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nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the nucleus. d. atomic mass. ____2.

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Chapter 6 What You'll Learn You will explain why elements in a group have similar properties Chapter 6 the periodic table chapter test a answer key. You will relate the group and period trends seen in the periodic table to the electron configuration of atoms. You will identify the s-, p-, d-, and f-blocks of the periodic table Chapter 6 the periodic ...

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Chapter 5 TEST: The Periodic Table
Multiple Choice Identify the choice that best completes the statement or answers the question ____ 1 The order of elements in the periodic table is based on a the number of protons in the nucleus b the electric charge

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